Preparation and antitubercular activity of lipophilic diamines and amino alcohols

Celso OR Júnior1, Mireille le Hyaric1, Cristiane F da Costa1, Taís A Corrêa1, Aline F Taveira1, Débora P Araújo1, Elaine FC Reis1, Maria Cristina S Lourenço2, Felipe RC Vicente2, Mauro V de Almeida2/+1

1Departamento de Química, Universidade Federal de Juiz de Fora, Juiz de Fora, MG, Brasil 2Instituto de Pesquisa Clínica Evandro Chagas-Fiocruz, Rio de Janeiro, RJ, Brasil

A series of diamines and amino alcohols derived from 1-dodecanol, 1-tetradecanol, 1,2-dodecanediol and 1,2-tetradecanediol were synthesized and tested for their antitubercular activity. Compounds 3, 8 and 9 were found to be the most active (MIC of 6.25 µg/mL). Nine other compounds displayed activity against Mycobacterium tuberculosis, with a MIC of 12.5 µg/mL.

Key words: tuberculosis - amino alcohols - diamines - lipophilicity

Tuberculosis (TB), a serious disease caused by Mycobacterium tuberculosis, is transmitted by air and can affect all organs of the body, especially the lungs. In case of infection, treatment basically consists of the combination of three medicines: rifampicin, isoniazid and pyrazinamide. If the medications are taken correctly, the chances of cure reach 95%. Interruption of treatment causes the emergence of drug resistant strains of TB, which is the main world-wide concern regarding the disease (WHO 2009). Multidrug resistant TB, which is resistant to isoniazid and rifampin, is a particularly dangerous form of TB, and its spread could lead to a new wave of a virtually incurable disease throughout the entire world (WHO 2009).

Several antitubercular compounds with different mechanisms of action have been synthesised in recent years. Their targets include fatty acid biosynthesis, protein synthesis, DNA-based processes, cell wall synthesis and energy metabolism pathways (Zhang et al. 2006, Janin 2007). Ethambutol, an amino alcohol first reported by Wilkinson et al. in the early 1960s, is one of the few long-known and reliable first-line antimycobacterial chemotherapeutic agents and acts as an inhibitor of cell wall biosynthesis (Janin 2007, Ralambomanana et al. 2008). Glycosylated amino alcohols and amines are known for their antimycobacterial activity (Katyar et al. 2005). We have previously described the synthesis and anti-TB activity of glycosylated derivatives of N-alkylated diamines or amino alcohols (De Almeida et al. 2007, Taveira et al. 2007). In the present work, we investigate the anti-TB activity of a series of N-alkylated diamines and amino alcohols obtained from fatty alcohols (Figure).

MATERIALS AND METHODS

Chemistry - Compounds 1-10 were prepared in two steps from 1-dodecanol or 1-tetradecanol using a methodology previously described (Reis et al. 2008). The alcohol was first treated with mesyl chloride in methylene chloride and pyridine to furnish the corresponding mesylate. The crude mesylates were treated with the amino alcohols (2-aminoethanol, 3-amino-1-propanol, diethanolamine or 2-amino-2-methyl-1-propanol) or diamines (ethylenediamine or 1,3-propanediamine) in ethanol at reflux to furnish the desired amino alcohols 1-8 and diamines 9 and 10 with a 42-98% yield in two steps. Compounds 11-22 were obtained with a 40-90% yield using a similar protocol starting from 1,2-dodecanediol or 1,2-tetradecanediol.

In vitro assay - The antitubercular activity of the tested compounds was determined by incorporating decreasing concentrations into a culture of M. tuberculosis H37Rv (ATCC 27294) in Middlebrook 7H9 agar medium using the Microplate Alamar Blue Assay (Collins & Franzblau 1997). MIC values represent means of three separate experiments. Ethambutol was used as the reference compound. LogP was calculated using ACDLABS software.

RESULTS AND DISCUSSION

The best results (MIC = 6.25 µg/mL) were obtained for compounds 3, 8 and 9, which were derived from dodecanol and tetradecanol (Table). Compounds 1 and 11, derived from diethanolamine, displayed the lowest activity of all the tested compounds, with a MIC = 50 µg/mL.

Comparing all the tested compounds individually, we could not find a clear correlation between lipophilicity (logP) (Table) and activity. However, when comparing compounds 11-22 prepared from diols, with compounds 1-10 obtained from monoalcohols, it appears that, with chain length and functional groups being the same, all diol derivatives were less lipophilic and less active than their monoalcohol analogues.
Comparing the length of the aliphatic chain, compounds 3 and 9 were the only compounds displaying better activity than their more lipophilic tetradecyl analogues 4 and 10 (MIC = 25.0 µg/mL). Compounds 2, 8, 12, 14, 18 and 20 were more efficient or as efficient as their less lipophilic dodecyl analogues 1, 7, 11, 13, 17 and 19.

Considering the presence of branching, the activities of amino alcohols 7, 8, 17 and 18 have been determined in previous work (Taveira et al. 2007, Reis et al. 2008), which showed that compounds 3 and 8 display low toxicity (0.5 µg/mL) while compounds 2, 4, 6, 14 and 16 do not display significant cytotoxicity.

In conclusion, this work describes the antitubercular evaluation of several diamines and amino alcohols derivatives containing an alkyl chain. Three of these compounds (3, 8 and 9) displayed good activity against M. tuberculosis. The results shown here, together with alcohols 15 and 16 (MIC = 25.0 µg/mL), but ethylenediamine derivative 20 (MIC = 25.0 µg/mL) was less active than its alcohol analogue 14 (MIC = 12.5 µg/mL).

Cytotoxicities of amino alcohols 2, 3, 4, 6, 8, 14 and 16 have been determined in previous work (Taveira et al. 2007, Reis et al. 2008), which showed that compounds 3 and 8 display low toxicity (0.5 µg/mL) while compounds 2, 4, 6, 14 and 16 do not display significant cytotoxicity.

In conclusion, this work describes the antitubercular evaluation of several diamines and amino alcohols derivatives containing an alkyl chain. Three of these compounds (3, 8 and 9) displayed good activity against M. tuberculosis. The results shown here, together with...
previously published data, provide evidence that N-alkylated diamines or amino alcohols could serve as good starting points in the search for new lead compounds.

REFERENCES


