

## Review Article

# Prenylation of Aromatic Secondary Metabolites: A New Frontier for Development of Novel Drugs

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### Abstract

Prenylation of aromatic secondary metabolites is an important process involved in the biosynthesis of many biologically active compounds in plants and animals. Many prenylated natural products have been shown to exhibit very good anti-tumor, anti-inflammatory and antimicrobial activities. The addition of prenyl side chain to an aromatic secondary metabolite often affects its pharmacological activity. The aim of this review is to examine the influence of the prenyl side chain on the pharmacological activities of some classes of aromatic secondary metabolites and the potential application of prenylation in the development of novel drugs

**Keywords:** Secondary metabolite, Prenylation, Aromatic, Ras antagonist, Novel drugs.

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## INTRODUCTION

Prenylation is a chemical or enzymatic addition of an isoprenoid side chain to another molecule. Prenylation of aromatic secondary metabolite is an important process involved in the biosynthesis of many biologically active compounds in plants and animals. Aromatic secondary metabolites containing prenyl side chains represent a rare class of natural products which are mainly found in plants belonging to the family of Rutaceae and to a lesser extent in the family of Apiaceae, Umbelliferae, Boraginaceae, Piperaceae, Compositae and Leguminosae [1]. A wide variety of pharmacological activities have been reported in plant secondary metabolites having prenyl residues.

The addition of a prenyl chain to an aromatic secondary metabolite often results in a derivative with improved or modified pharmacological activity. These "hybrid" natural products represent nowadays a new frontier for the

development of novel drugs, in particular as antimicrobial, anti-oxidant, anti-inflammatory and anti-cancer agents [1]. The prenyl side chains are of different carbon length (Figure 1). Four types of prenyl side chain have been identified based on the size of the carbon; C5 (isopentenyl), C10 (geranyl), C15 (farnesyl) and C20 (geranylgeranyl). Compounds having the isopentenyl and the geranyl side chains are more abundant in nature while those with farnesyl and geranylgeranyl are rare. The prenyl skeleton is attached to the aromatic compound directly or through another atom such as oxygen, nitrogen or sulphur. The carbon and oxy prenylated metabolites are quite abundant while the later two are less common.

The biosynthesis of prenylated aromatic secondary metabolites involves the coupling of products from the shikimic acid and the mevalonic acid pathways. The latter produces the prenyl group while the former produces the

aromatic compound. The two products are linked together by prenyl transferase enzymes [2]. Many prenylated natural products have been isolated long ago, but it was only in the last decade that this class of compounds received tremendous research attention particularly, in the aspect of their pharmacological activity evaluation. Many review papers published on this subject area have focused on the isolation, classification and general pharmacology of prenylated natural products. In this review however, we have examined the influence of the prenyl group on the pharmacological activities of some classes of aromatic secondary metabolites and the potential application of prenylation in development of novel drugs.

## INFLUENCE OF THE PRENYL SIDE CHAIN ON THE PHARMACOLOGICAL ACTIVITY

### Prenylated acridones

Prenylated acridones are mainly found in citrus plant family of Rutaceae. They have been shown to possess various pharmacological activities, such as cytotoxicity [3,4], antiviral [5] and antimalaria activities [6,7]. Itoigawa and co workers [8] evaluated 17 acridone alkaloids isolated from plants of the genera Citrus,

Glycosmis and Severinia (Rutaceae) for their anti-tumor-promoting activity in the short-term in vitro assay of Epstein-Barr virus early antigen (EBV-EA) activation. They discovered that the prenylated acridones **3-10** (Figure 2) have remarkably more potent activities compared to the non prenylated compounds with the same basic structure. Glycocitrine-II (**3**) and Omethylglycocitrine- II (**5**) exhibited more potent inhibitory activity ( $IC_{50}$  of 280 and 281 mol ratio/32 pmol TPA respectively) when compared to 1,3-Dihydroxy-10-methylacridone (**1**) which do not have prenyl side chain ( $IC_{50}$  of 368 mol ratio/32 pmol TPA).

Similarly the prenylated pyranoacridones (**6-9**) exhibited more potent antitumor promoting activity ( $IC_{50}$  276–302 mol ratio/32 pmol TPA), compared to des-N-methylnoracronycine (**2**) which has the same basic structure but lack the prenyl side chain ( $IC_{50}$  437 mol ratio/32 pmol TPA). Based on the results of this investigation, the group synthesized 1,3-dihydroxy-10-methyl-2,4-diprenylacridone (**10**) a novel compound which exhibited a marked inhibitory effect on mouse skin tumor promotion in an *in vivo* two-stage carcinogenesis test (52% reduction in incidence of papiloma in 20 weeks). These findings

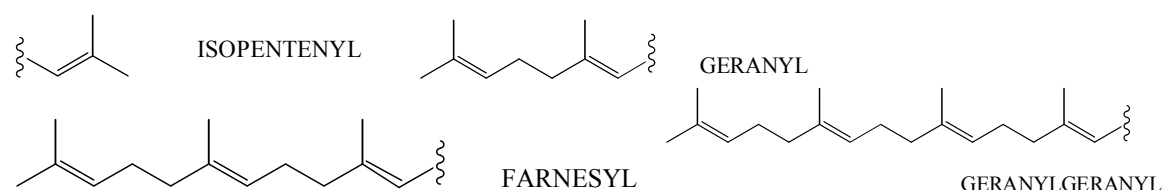


Figure 1: The prenyl side chains

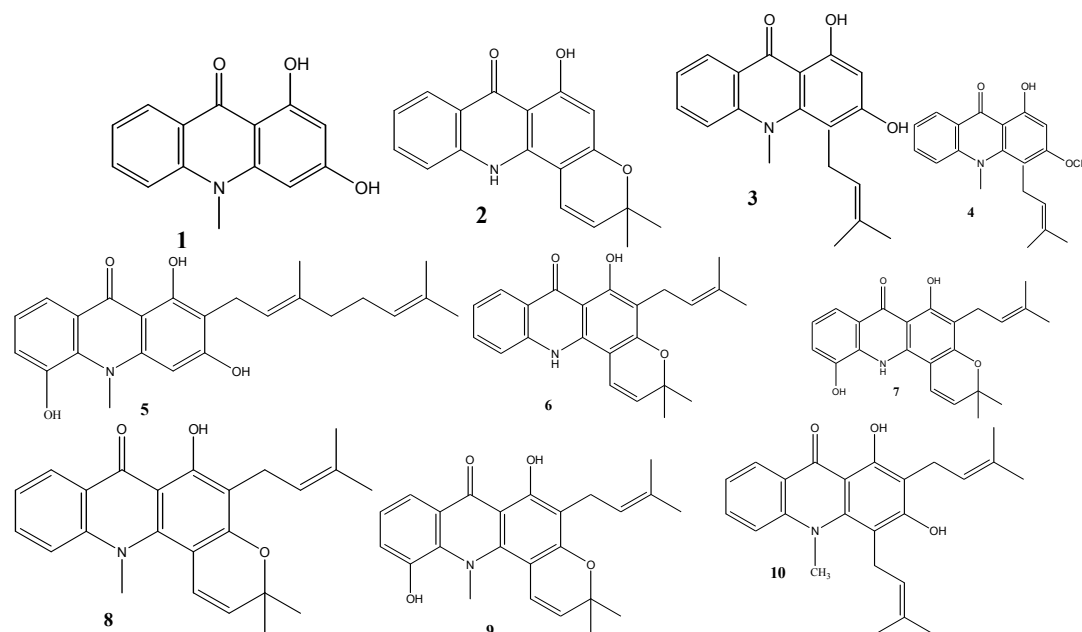


Figure 2: Prenylated and non-prenylated acridones

indicate that the prenyl group is an important structural feature for good anti tumor promoting activity of the acridone molecule and can be exploited in new drug development.

### Prenylated coumarins

This class of compounds is commonly found in citrus plants, family Rutaceae. They frequently occur as oxyprenylated metabolites. The oxyprenylated coumarins possess good anti-inflammatory and anticancer properties. The first oxyprenylated secondary metabolite to be reported in literature is a coumarin, auraptene (**11**), isolated in 1930 from *Citrus aurantium L.* (rutaceae) and structurally characterized by Kariyone and Matsuno [9]. It was later discovered to be present in other plants like *Zanthoxylum schinifolium* [10] and *Paliurus ramosissimus* [11]. Auraptene was shown to be effective in preventing the chemical carcinogenesis in various rodent models including skin [12], tongue [13], liver [14], and colon carcinogenesis [15]. Another natural occurring prenyloxy coumarin is umbelliprenin (7-trans, trans-farnesyloxy coumarin) (**12**), isolated from *Ferula persica* [16]. It was shown to possess antioxidant, anti-inflammatory and lipoxygenase inhibitory activities [17]. Based on the foregoing, the effects of prenyl group on anticancer and anti-inflammatory activities of coumarins were investigated. Devji and Co-workers [18] synthesized a series of prenylated and non prenylated hydroxycoumarin derivatives and evaluated their activity against human pancreatic PANC-1 cancer cells under nutrient-deprived conditions. The results of the investigation showed that the prenylated coumarins **18 - 22** (Figure 3) exhibit higher cytotoxic activity against the PANC-1 cancer cells than **17**, while **13-16** do not show any appreciable activity. The results also showed that the longer the prenyl side chain the higher the cytotoxic activity. This study led to the discovery of a novel geranylgeranylated hydroxycoumarin (**21**) which exhibited 100 % preferential cytotoxicity against PANC-1 cells under nutrient-deprived medium at 6.25  $\mu$ M making this compound a powerful new lead structure for the development of novel anti-cancer agents. In another related work conducted [19], O-prenylated derivatives involving farnesyloxy, geranyloxy and isopentenylloxy substituents at positions 3, 4, 5, 6, 7 and 8 of coumarin ring were synthesized and their lipoxygenase inhibitory activities together with the SAR studies were evaluated. The results obtained from these experiments showed that among the three groups of synthetic O-prenylated coumarins, farnesyloxy derivatives

displayed the best inhibitory activity against soybean 15-lipoxygenase ( $IC_{50}$  : 1.7 to 5.8  $\mu$ M) while the isopentenyl derivatives displayed the least ( $IC_{50}$  : 39.8 to 69.4  $\mu$ M). The results further showed that the Lipoxygenase inhibitory potential of the coumarins is influenced by the presence and length of prenyl side chain and it decreases by 8 to 50 fold from farnesyl to isopentenyl substituents.

### Prenylated flavonoids

Watjen and co-workers [20] analyzed the antioxidative potential and cytotoxic effects of two prenylflavonoids (Figure 4), licoflavone C (**24**) and isobavachin (**26**) in comparison to their non prenylated analogues, apigenin (**23**), liquiritigen (**25**) and apigenin glucoside (**27**). They discovered that compounds **24** and **26** showed good cytotoxic activity against C6 glioma and H4IIE hepatoma cells with  $IC_{50}$  of  $42 \pm 5$  and  $96 \pm 19$   $\mu$ mol/L respectively. In contrast, compounds **23**, **25** and **26** did not show any significant cytotoxic activity against the two cancer cell lines. This indicates that the insertion of prenyl group (isopentenyl) at position 8 enhances the cytotoxic activity of apigenin and liquiritigen. This property may probably be extended to other flavonoids. Further research is needed to confirm this.

### Prenylated chalcones

Prenylated chalcones are commonly found in hops (cannabaceae) and few other plants. This class of compounds exhibit good antioxidant activity. The ability of some prenylated and non prenylated chalcones to inhibit the *in vitro* oxidation of human low density lipoprotein (LDL) was investigated by Miranda and Co workers [21]. The oxidation of LDL was assessed by the formation of conjugated dienes and thiobarbituric acid reactive substances (TBARS) and Tryptophan fluorescence. They found that the prenylated chalcones (**29**, **30**) inhibited the oxidation of LDL (50  $\mu$ g protein/ml) at concentration of 5 and 25  $\mu$ M while the non prenylated chalcone (**28**) (Figure 5) exerted pro-oxidant activity on LDL. This is an indication that the presence of prenyl group significantly enhances the antioxidant activity of hydroxylated chalcones.

### Prenylated cinnamic acids

Artepillin C (**31**) and 4 geranyloxyferulic acid (**34**) are prenylated cinnamic acid derivatives with interesting and valuable pharmacological properties. Artepillin C (3,5-diprenyl-4-hydroxycinnamic acid) was isolated from green

Brazilian propolis which is produced from exudates of *Baccharis dracunculifolia* [22,23]. Artepilli C possesses very good anti-inflammatory property [24]. Paulino and Co workers [24] investigated the anti-inflammatory effects, absorption, and bioavailability of artepillin C in mice. The mice were subjected to: carrageenan-induced paw oedema (300 µg/paw), carrageenan-induced peritonitis, and prostaglandin E2 determination. In addition, *in vitro*

nitric oxide production by RAW 264.7 cells and NF-κB activity in HEK 293 cells were also measured. The group reported that *in vivo* Artepillin C produced a maximal inhibition of 38% after 360 min on paw oedema, decreased the number of neutrophils during peritonitis (IC<sub>50</sub>: 0.9 (range 0.5 – 1.4) mg/kg) as well as decreased prostaglandin E2 by 29±3% and 58±5% at 1 and 10 mg/kg, respectively, with a mean ID<sub>50</sub> of 8.5

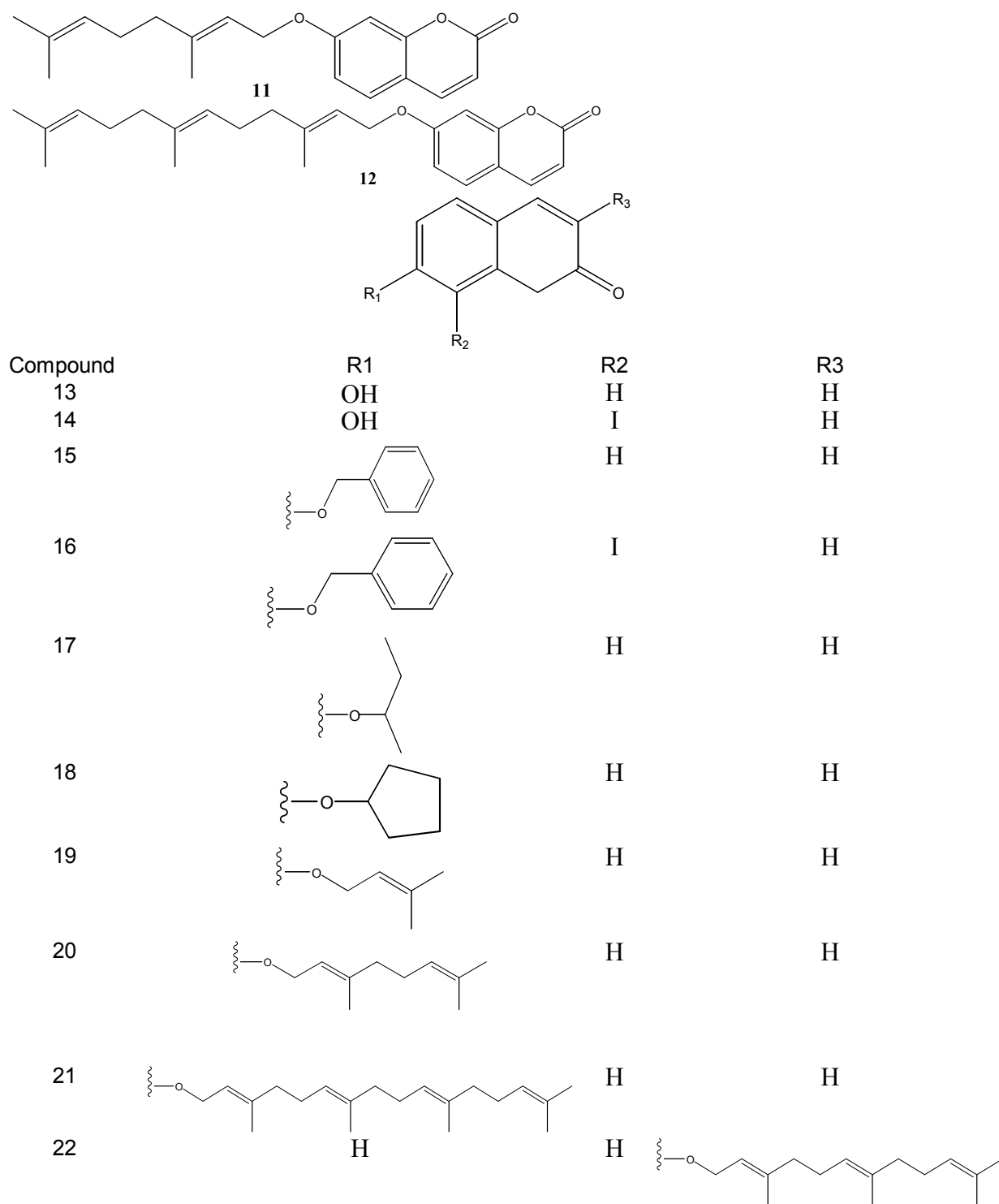
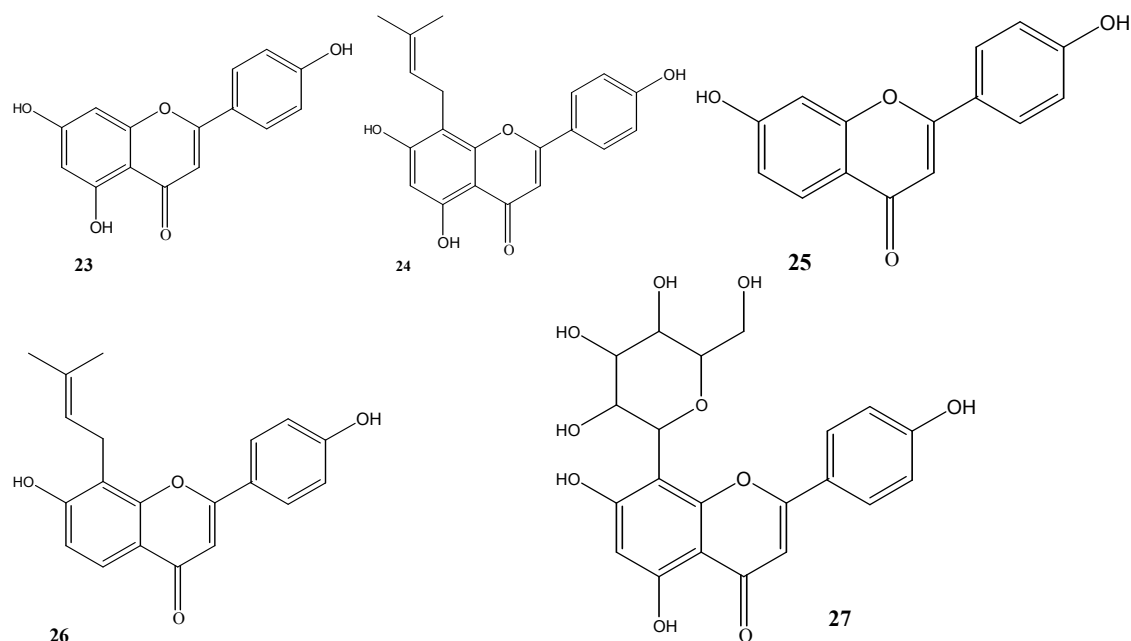
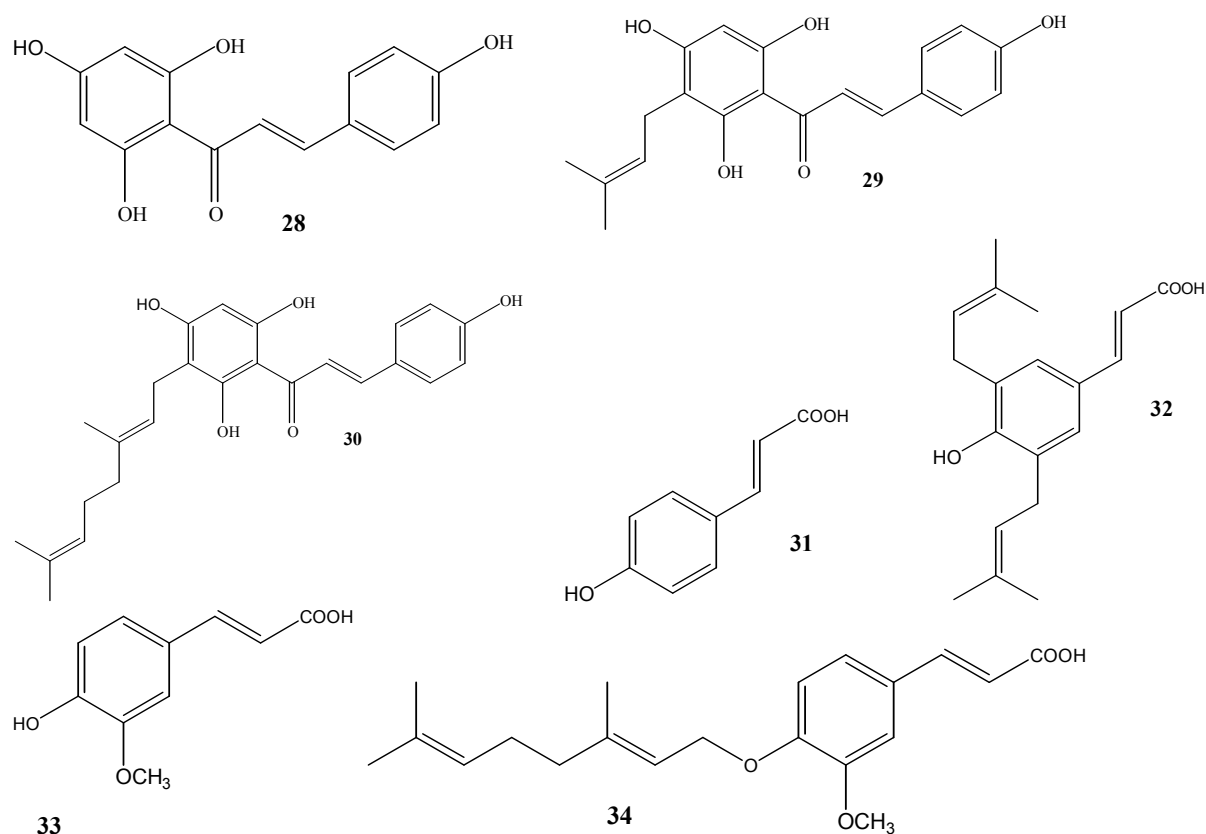


Figure 3: Structure of prenylated and non-prenylated coumarins



**Figure 4:** Structures of prenylated and non-prenylated flavonoids



**Figure 5:** Structures of Prenylated and non prenylated chalcones (28-30) and cinnamic acids (31-34)

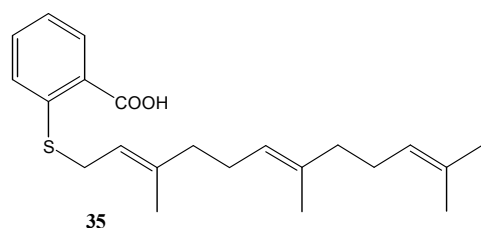
(range 8.0 – 8.7)mg/kg). Similarly, in *in vitro* models, it decreased nitric oxide production by RAW264.7 cells with a mean  $IC_{50}$  of 8.5 (range 7.8 – 9.2)  $\mu$ M. This high anti-inflammatory activity displayed by this compound could probably be due to the presence of diprenyl group attached at

position 3 and 5 of the molecule because the non prenylated 4-hydroxy cinnamic acid scaffold (**31**) (Figure 5) has been shown to possess only mild to moderate anti-inflammatory activity [25]. In addition to its anti-inflammatory property, artemillin C has also been shown to exhibit

remarkable antimicrobial [26] anticancer [27-29] and antioxidant activities [30,31], which may also be related to the presence of the diprenyl group. 4-Geranyloxyferulic acid (**34**) (Figure 5) was isolated from the bark of *Acronychia baueri* Schott, an Australian plant belonging to the family of Rutaceae [32]. This compound was shown to display good antitumor properties, particularly against colon related cancers [33-35]. This valuable pharmacological property has not been reported in the unprenylated ferulic acid (**33**) (Figure 5). This indicates that the presence of geranyloxy residue may have been partly responsible for the good anticancer property of the molecule.

### MECHANISM OF ACTION

One possible mechanism by which a prenyl group may enhance or modify the pharmacological activity of aromatic secondary metabolite is through improvement of lipophilicity. The introduction of a prenyl side chain into a molecule increases its lipophilicity which could modify its pharmacological activity through enhanced access, affinity and interaction with the lipophilic membrane [36]. Another possible mechanism is the inhibition of ras signal transduction. Ras proteins constitute an important part of intracellular signal transduction system. Ras proteins are activated by attachment of farnesyl moiety. Activated RAS proteins are involved in regulating many cell functions, such as proliferation, differentiation and inflammation. RAS mutations are found in approximately one-third of human cancers, most especially in pancreatic and colorectal carcinomas [37]. Some prenylated aromatic compounds particularly, farnesyl thiosalicylic acid (**35**) (Figure 6) and its analogues were shown to interrupt the Ras signal cascade by interfering



**Figure 6:** Structure of farnesylthiosalicylic acid

with the binding of RAS to its plasma membrane anchor protein [38-39]. Besides these two mechanisms, the prenyl moiety could exert its activity by allowing indigenous molecules to identify specific but still unknown biological targets in human and animals [40]. Further

studies are, however, needed to understand the detailed mechanisms by which prenyl group influences pharmacological activity.

### FARNESYL THIOSALICYLIC ACID - AN EXAMPLE OF PRENYLATION IN DRUG DEVELOPMENT

Salirasib (*S-trans, trans*-farnesylthiosalicylic acid, FTS, Concordia Pharmaceuticals, Ft. Lauderdale, FL) is a molecule in which the hydroxyl group of salicylic acid was replaced with a *S*-farnesyl group. FTS interferes with the interactions of Ras with its binding site which results in dislodgment of Ras from the cell membrane, thus preventing Ras activation and signal transduction cascade. FTS has been shown to inhibit the *in vitro* Ras dependent proliferation of different cancer cell lines including glioblastoma [41], pancreatic carcinomas [42], ovarian carcinomas [43] colon carcinomas [44] and prostate cancer cells [45]. FTS is currently in clinical trial. Results from the phase 1 clinical trial showed that it is effective, relatively safe and pharmacokinetics data suggest the possibility of this molecule emerging as the first available orally active Ras antagonist [46]. It is now evident that the insertion of prenyl group into an aromatic compound is a viable means through which new drugs with improved pharmacological activities could be developed. The challenge now is to extend such studies to cover wide variety of aromatic compounds including the existing drugs that are already in use.

### CONCLUSION

Quite a large number of prenylated aromatic metabolites have been isolated as natural products while few were synthesized. Many studies about their pharmacological activities have been carried out in the last decade. Results from these investigations have clearly demonstrated how the presence of prenyl group improved their pharmacological activities; thus indicating the possibility of aromatic secondary metabolites as potential source of novel drugs for the treatment of diseases such as cancer, inflammation and microbial infections. The discovery of farnesyl thiosalicylic acid as Ras antagonist, further buttress the capacity of prenylation of aromatic compounds in development of new drugs. It is anticipated that future research in this area will cover a wider range of aromatic compounds besides the alkaloids and phenylpropanoids. Moreover, further studies are needed to understand the detailed mechanisms by which prenyl group influence pharmacological activity. This could

provide a basis for rapid development of drug for the benefit of humanity.

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