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Statistical physical and nutrient optimization of bioplastic polyhydroxybutyrate production by *Cupriavidus necator*

A. Aramvash · Z. Akbari Shahabi · S. Dashti Aghjeh · M. D. Ghafari

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Abstract Polyhydroxyalkanoates are biodegradable polymer materials that accumulate in numerous bacteria. The polyhydroxybutyrate is the most common type of polyhydroxyalkanoates, which potentially serves as precursor for bioplastic production. The most extensively studied polyhydroxybutyrate producing bacteria is Cupriavidus necator due to its capability to accumulate large amounts of this biopolymer in simple culture medium. Accumulation of polyhydroxyalkanoates granules in the cytoplasm of C. necator significantly depended on pH, aeration, carbon sources, nitrogen sources, and minerals in the culture medium. In the present study, the effect of both nutritional and physical variables on polyhydroxybutyrate production was investigated in order to optimize these conditions. At first, on the basis of one-factor-at-a-time experiments, fructose and ammonium chloride were found to be the most suitable sources of carbon and nitrogen for biopolymer production. Then the most significant factors affecting granules accumulation were recognized as fructose, agitation speed, KH₂PO₄, and initial pH using the Plackett-Burman and central composite design. ANOVA analysis showed significant interaction between fructose and agitation speed. After optimization of the medium,

A. Aramvash (🖂) · S. Dashti Aghjeh

Department of Bioscience and Biotechnology, Malek-Ashtar University of Technology, P.O. Box 16765-3454, Tehran, Iran e-mail: Aramvash@ibb.ut.ac.ir

Z. Akbari Shahabi

Biotechnology Group, Department of Chemical Engineering, Iran University of Science and Technology, Tehran, Iran

M. D. Ghafari Microbiology Group, Department of Biology, Shahed University, Tehran, Iran compositions for polyhydroxybutyrate production were determined as follows: fructose 35 g/L, KH_2PO_4 1.75 g/L, $MgSO_4 \cdot 7H_2O$ 1.2 g/L, citric acid 1.7 g/L, trace element 10 mL/L, initial pH = 7, and agitation speed 175 rpm. Under this optimal culture conditions, the maximum yield of PHB was 7.48 g/L. The present strategies included in this study could be used for PHB production by this bacterium. These results are the highest values of PHB ever obtained from batch culture of *C. necator* reported so far.

Keywords Biopolymer · Optimization · Plackett– Burman · Polyhydroxyalkanoates · Response surface methodology

Introduction

A large body of evidence has shown that plastics can persist in the environment for hundreds of years often as harmful microscopic particles that are known to be dangerous to the environment, especially for marine ecosystems (Galgani et al. 2000; Mumtaz et al. 2010). They are also produced from petrochemical sources that may be depleted and take millions of years to be renewed. Recently, these negative environmental impacts have increased concern of the global community around the world (Gregory 2009). However, there are several attempts to remove toxicity and pollution of water and land (Colak et al. 2009, 2011); some research is focusing increasingly on the development of environmental friendly methods (Gupta et al. 2013) for the production of eco-friendly materials that combine the desired functionality during use and rapid degradation after disposal as an alternative to conventional nondegradable materials. Bioplastics fit this context and are generated from renewable natural sources



and are often biodegradable and nontoxic (Poirier 1999; Flieger et al. 2003). They can be produced by biological systems or chemically synthesized from biological materials (Flieger et al. 2003).

Polyhydroxyalkanoates (PHAs) are thermoplastic biopolymers that accumulate in wide range of microorganisms under stress conditions (Chen 2005; Khanna and Srivastava 2005). In recent years, PHAs have attracted increasing attention because they could be substitutes for conventional plastics (such as polyolefins), which can reduce petroleum consumption and alleviate environmental impact caused by plastic waste (Ivanov et al. 2014). These bioplastics have application in agriculture, medicine, pharmaceutical, and packaging (Chen and Qiong 2005; Wang et al. 2005). The most widely produced microbial PHA is polyhydroxybutyrate (PHB). *Cupriavidus necator* has remarkable capacity for accumulating PHB up to 90 % of the cellular dry weight under phosphorous or nitrogen limitation (Liu et al. 2005; Shang et al. 2003).

It is obvious that the optimization of media parameters plays an important role in increasing the production of any metabolite in setting up of any microbial fermentation process. The one-factor-at-a-time method optimizes medium by changing one factor of the medium and keeps all other parameters constant. However, this method is timeconsuming, and interaction among the medium component cannot be studied (Mu et al. 2009). The limitation of such method can be avoided by using statistical experimental design, a powerful and useful tool, which can allow explaining interactions between the different variables and decreasing the process variability (Frey et al. 2003). The most popular choices are the Plackett-Burman design (PBD) and the central composite design (CCD), along with the response surface analysis (Liu et al. 2005; Xu et al. 2008). The response surface methodology (RSM) is a kind of statistical technique for designing experiments, evaluating the relative significance of several independent variables, and determining the optimum conditions for desirable responses (Anderson-Cook et al. 2009). There are some examples of successful optimization of medium for microbial production of PHB in various microorganisms. These studies take into account the effect of different media components on maximizing the productivity of PHB using various nitrogen sources (i.e., urea and NH₄Cl) and carbon sources (i.e., sucrose, glucose, acetate, malt, soya, sesame, molasses, bagasse, and gluconic acid) (Md Din et al. 2014; Arun et al. 2006; Sandhya et al. 2013; Wei et al. 2011). KH₂PO₄, Na₂HPO₄, and MgSO₄·7H₂O and physical factors like temperature, pH, and agitation speed which playing significant role in PHB accumulation were also optimized in some microorganisms using RSM for enhanced PHB production (Khanna and Srivastava 2005; Cavalheiro, et al. 2009; De Almeida et al. 2010). However,



there are few reports, which focused on the effect of physical process variables along with nutrient constituents on PHB production (Khosravi Darani et al. 2004). In this study at first, the one-factor-at-a-time experiments were used to select the optimal carbon and nitrogen sources. Then, Plackett-Burman design was used to identify the most important factors on PHB production, which shows physical variables (i.e., agitation speed) are as important as medium components, because the redox state of the cells is known to affect the synthesis of PHB (de Almeida et al. 2010). Moreover, CCD and the analysis of response surface methodology (RSM) were carried out to optimize effective factors including fructose, KH₂PO₄, initial pH, and agitation speed for enhanced PHB production in batch cultivation of C. necator ATCC 17699. This study was conducted during September 2013 to April 2014 at Laboratory of Biochemistry, Department of Bioscience and Biotechnology, Malek-Ashtar University of Technology, Tehran, Iran.

Materials and methods

Microorganism strain and inoculum preparation

Cupriavidus necator ATCC 17699 was obtained from the Persian Type Culture Collection for the production of PHB. The PHB-producing capability of the organism was confirmed by Sudan Black staining method (Burdon 1946). The bacteria were maintained on LB agar slants at 4 °C. For inoculums preparation, the cells were grown on a LB medium for 10 h at 30 °C on a shaker at 200 rpm; stock cultures were maintained at -20 °C in 5-mL vials containing 4 mL of basal medium and 1 mL of glycerol solution until use.

Culture condition

The composition of the basal mineral salt medium (MSM) used in this study was as follow: $(NH_4)_2SO_4$ 2.0 g/L, KH_2PO_4 1.54 g/L, $MgSO_4 \cdot 7H_2O$ 1.2 g/L, Citric acid 1.7 g/L, trace metal solution 10 mL/L (Kim et al. 1994a). The trace element stock solution composed of $ZnSO_4 \cdot 7H_2O$ 2.25 mg/L, $FeSO_4 \cdot 7H_2O$ 10 mg/L, $CaCl_2(2H_2O)$ 2 mg/L, $Na_2B_4O_7(7H_2O)$ 0.23 mg/L, $(NH_4)_6Mo_7O_{24}$ 0.1 mg/L, $CuSO_4 \cdot (5H_2O)$ 1 mg/L, $MnSO_4(5H_2O)$ 0.6 mg/L, and HCl 35 % 10 mL/L (Kim et al. 1994b). Glucose was used as carbon source in concentration of 20 g/L for production media and inoculums development. (Glucose was used for initial studies; however, after some experiments, fructose was used as carbon sources for the following experiments). The initial pH was adjusted to 7.0.

Comparison of different carbon and nitrogen sources was carried out using the media same as described above.

However, for comparison, different carbon (20 g/L) and nitrogen (2 g/L) sources at constant concentration were taken.

Analysis of culture biomass and PHB production

The effect of different incubation time on growth and PHB production was carried out using MSM supplemented with 2 % glucose. 10 % inoculums were added into 100 mL of production medium and incubated at 200 rpm, 30 °C for different time points (2–70 h). Samples were withdrawn every 2 h and analyzed for biomass and PHB. These experiments showed that 48 h is sufficient for harvesting cells, because after 48 h, the cells reach at steady state phase, and at this time, there is maximum PHB production.

Effect of different carbon sources on growth and PHB production

The experiments were performed one factor at a time. The effect of different carbon sources on growth and PHB production of culture was carried out using MSM media supplemented with different carbon sources (i.e., fructose, glucose, lactic acid, arabinose, sucrose, and lactose) at 20 g/L. Inoculums were prepared as described above. The flasks were incubated at 200 rpm at 30 °C. Biomass and PHB were estimated in the culture broth after 48 h.

Effect of different nitrogen sources on growth and PHB production

These experiments were done in MSM containing fructose (20 g/L) as carbon source and different nitrogen sources (2 g/L) (i.e., ammonium chloride, ammonium sulfate, peptone, tryptone, yeast extract, and urea). The flasks were incubated at 30 $^{\circ}$ C and 200 rpm. Biomass and PHB were estimated in cell broth after 48 h.

Growth and cell dry weight determination

Cell growth was monitored over time by the measurements of optical density at 600 nm (OD_{600}) using jenway 6310 (Bibby Scientific Limited, Staffordshire, UK).

Cell dry weight determination was done by weighing the dry cell mass obtained as follows. 10 ml culture samples were centrifuged at 10,000 rpm for 15 min at 4 °C. The pellet was resuspended in distilled water (10 mL) and centrifuged again for washing. The washed cells were dried at 90 °C for 24 h in a hot air oven. The drying was repeated until constant weight was obtained (Jung et al. 2000).

Detection of PHB

For determining PHB amount, 10 mL of 98 % sulfuric acid was added to bacterial pellet at 90 °C for 1 h; PHB crystals in cytoplasm of bacteria were converted into crotonic acid. The absorbance of the solution was measured at 235 nm in a UV spectrophotometer against sulfuric acid as blank. The amounts of PHB per gram dry weight of bacterial cells were determined using a standard curve of PHB according to Singh and Parmar (2011).

Plackett–Burman design and statistical analysis for optimization

Optimization of medium components for PHB production was performed in two stages. At the first stage, the components that have significant effect on PHB production were identified. At the second stage, the optimum values of these components were determined. A total of seven factors were tested, which were fructose, KH₂PO₄, initial pH, ammonium chloride, agitation speed, inoculum size, and trace element solution, with each being represented at two levels, high (+1) and low (-1). A design of 16 experiments was formulated for seven factors using the software. Concentration range for the variables was decided on the basis of other reports for PHB production by C. necator (Tripathi et al. 2013; Khanna and Srivastava 2005). The experiments were done in flasks containing 100 mL media at 200 rpm for 48 h in duplicate. The factors with confidence level above 95 % are considered the most significant factor that affects PHB production. Four factors (fructose, KH₂PO₄, agitation speed, and initial pH), screened from Plackett-Burman design, were studied for determining their optimum concentration values for biomass and PHB production.

Response surface methodology

Central composite design using Design Expert (version 7.0.0) software (Stat-Ease Corporation, USA) was used for optimization of the media components, which were found to be important for PHB production according to screening results, based on Plackett–Burman, rest of the factors being kept at a constant level. In this study, the experimental plan consisted of 30 experiments was formulated using the Design Expert software. Experiments were conducted in 250-mL flasks containing 50 mL of media prepared according to the design. The incubator was set at 30 °C. Responses studied were biomass (g/L) and PHB (g/L); at the end of 48 h, three-dimensional (3D) plots were generated to understand the interaction of various factors and then used to find the optimized concentration of the media components affecting the response. A special feature of the



software point prediction was used to confirm the aboveobtained optimized values.

Results and discussion

Initial culture and PHB production capacity

Initial cultivation of *C. necator* was performed at 30 °C and 200 rpm by using the media described in the "Materials and methods." Initial pH of the media was adjusted to 7.0. Maximum biomass obtained was 1.375 g/L with PHB content of 73 % at 48 h (Fig. 1). The pH value measured at the end of the cultivation was 7.55 compared with initial value of 7.0. This result shows the medium has relatively strong buffering capacity compared with Khanna and Srivastava (2005) experiment that shows high decrease in pH from 7 to 5.26 after 48 h. This difference may be due to lower level of carbon source that we used (20 g/L) and usage of glucose instead of fructose in initial study. Owen et al. (1991) reported that fructose consumption products lower the pH of media more than other source of carbohydrates.

The effect of different carbon sources on growth and PHB production

Based on the media used above, a comparative study was carried out by growing the culture in different carbon sources. Finding which carbon source is suitable for *C. necator* for maximum PHB production gives us insight on using which cost-effective carbon sources are suitable for this strain. Thus, six different carbon sources were selected that coverage the constituents of different cost-effective carbon sources. The results presented in Table 1 indicate



Fig. 1 Biomass, pH, and PHB content for initial shake flask culture. (*red filled circle*) PHB content (%); (*blue filled square*) Biomass (g/L); (*green triangle*) pH



 Table 1 Biomass and PHB production by growing the culture in different carbon sources after 48 h

Carbon source	Biomass (g/L)	PHB (g/L)	Final pH	
L-Arabinose	0.96	0.1	7.5	
Lactic acid	2.18	0.66	8.2	
Glucose	5.14	1.38	8.2	
Fructose	4.76	2.16	8.2	
Sucrose	3.28	0.29	7.93	
Lactose	2.92	0.27	8.28	

that among the six different combinations of cultures checked for PHB production, maximum PHB production was observed for fructose followed by glucose, lactic acid, sucrose, lactose, and arabinose. Table 1 also shows a maximum biomass of 5.14 g/L was obtained with 20 g/L of glucose. Although the highest growth rate was obtained from glucose, as compared to fructose, the PHB content was less. Because biomass production in glucose and fructose is near and PHB production in fructose is higher, fructose was selected as carbon source for further studies. Minimum biomass and PHB were obtained with 20 g/L-Arabinose. Previous data in the bioreactor showed that both growth and PHB production rates were slower on pentoses than on hexoses (Bertrand et al. 1990). Naturally occurring strains of C. necator have been reported to utilize only fructose as carbon source (Raje and Srivastava 1998). Therefore, for further optimization studies, fructose was used as carbon source.

Analysis of final pH confirms strong buffering capacity of our culture. In all experiments with different carbon sources, the cell growth and PHB production occur without changing pH significantly.

The effect of different nitrogen sources on growth and PHB production

Production of PHB by C. necator on different nitrogen sources was followed using MSM with fructose as a carbon source. Six different nitrogen sources were selected at concentration of 2 g/L. The results are shown in Table 2 and indicate that ammonium chloride produced the highest amount of PHB followed by yeast extract. The PHB content was relatively the same when urea, peptone, ammonium sulfate, and tryptone were used. However, the final biomass is lower in the presence of ammonium sulfate. This in contrast to the study of Koutinas et al. (2007) reported that ammonium sulfate is the best nitrogen source for the PHB production of C. necator. Also, other studies showed that Azotobacter vinelandii UWD strain PHB production increases in complex nitrogen sources, which is not applicable here (Page 1992). Because of these results, the ammonium chloride salt was used as a nitrogen source

 Table 2 Biomass and PHB production obtained with different nitrogen sources after 48 h

Nitrogen source	Biomass (g/L)	PHB (g/L)	Final pH	
Peptone	3.24	0.36	8.32	
Tryptone	3.34	0.31	8.25	
Yeast extract	4.1	0.59	8.1	
Urea	3.6	0.303	8.245	
$(NH_4)_2SO_4$	2.04	0.35	8.005	
NH ₄ Cl	5.6	0.66	5.185	

in the following studies. According to the final pH, pH values in this experiment decrease significantly, but it seems that it did not greatly affect the growth or production of PHB by the *C. necator*. This result is also in contrast with the results of Beaulieu et al. (1995) and Repaske (1962) who reported that (in the absence of pH control) at an ammonium concentration higher than 0.5 g/L, the growth of *C. necator* rapidly stopped when the pH dropped below 5.4.

Plackett-Burman design

Prior to optimization, it is necessary to use Plackett-Burman experiments that highlighted the importance of optimizing culture variables in attaining higher PHB production rate. In order to select significant variables for PHB production, a variety of factors including nutrients (fructose, KH₂PO₄, NH₄Cl, and trace element), physical (pH and agitation speed), and cultivation parameters (inoculum size) were tested and identified via Plackett-Burman design experiment (Table 3). The software used in the study was Design Expert 7.0.0 (Stat-Ease, Inc., Minneapolis, MN, USA). Seven factors were screened in sixteen combinations, and all the experiments were performed in duplicate, and average of the observation was used as the response of the design. Table 4 shows the distribution of the factors in the experiment according to the design expert software and the response in the study. Table 5 presents p values for the effect of medium components on biomass and PHB production. It is apparent from the table that pH, agitation speed, fructose, and KH₂PO₄ showed significant influence on PHB accumulation when considered as individual process variable (p < 0.05). Of the seven factors (A-F), except inoculums percentage, ammonium chloride, and trace element concentration, all others were taken for optimization, since the *p* value of them is greater than 0.05. On analysis of regression coefficient of sixteen medium components, fructose and agitation speed showed positive effect for PHB production, whereas KH₂PO₄ and initial pH showed negative effect in the tested range. From Table 4, it can also be seen that PHB production was a maximum when KH₂PO₄ was minimum (Experiments 7), and

 Table 3 Different variables at different levels used for Plackett– Burman design

Factor	Lower	Higher
Fructose (g/L)	5	25
NH ₄ Cl (g/L)	1	6
KH_2PO_4 (g/L)	0.5	3
Inoculum (%)	5	10
Trace element (mL/L)	5	15
Agitation Speed (rpm)	100	250
Initial pH	6	8

fructose and agitation speed are maximum. In this connection, Cavaillé et al. (2013) reported that PHB accumulation was enhanced when growth was restricted due to unavailability of phosphorus. However, their contribution to biomass production is also important. However, the study presented here succeeded in producing 31.6 % PHB, regarding to the cell dry weight. In this context, PHB content in *C. necator* cells varied from 3.6 to 30.5 % PHB. This means that the method used herein is a successful method for increasing PHB content.

Optimization by response surface methodology

The next step in the optimization of the medium was to determine the optimum levels of significant variables for PHB production. For this purpose, CCD was used to optimize the most significant factors affecting PHB production obtained from Plackett–Burman. Response surface methodology is a successive, exploratory approach to establish the relationship between more than one variable with the obtained responses. The analysis develops a model by fitting the experimental data in a generalized smooth curve, from which a specific predicted response could be calculated (Elibol 2004).

Base on the Plackett–Burman design, the significant variables utilized were as follows: fructose, KH_2PO_4 , agitation speed, and initial pH. The experiments conducted in the present study were targeted toward the construction of a quadratic model consisting of 30 trials. The minimum and maximum ranges of the variables were used, and the full experimental plan with regard to their values in actual form is provided in Table 6. The PHB production in each experiment was the average of the duplicates. The results were analyzed by Design Expert Software, and PHB production and biomass can be expressed in terms of the following equations:

$$\begin{split} \text{PHB} &= +3.10 + 1.88\text{A} - 0.25\text{B} + 0.26\text{C} + 0.47\text{D} \\ &\quad + 0.27\text{AB} - 0.12\text{AC} + 0.62\text{AD} - 0.090\text{BC} \\ &\quad + 0.47\text{BD} - 0.049\text{CD} + 0.21\text{A}^2 - 0.27\text{B}^2 \\ &\quad - 0.33\text{C}^2 - 0.43\text{D}^2 \end{split}$$



Table 4 Experimental design and response of Plackett-Burman study

Experiment No.	Fructose (g/L)	NH ₄ Cl (g/L)	KH ₂ PO ₄ (g/L)	Inoculum (%)	Trace element (mL/L)	Agitation speed (rpm)	Initial pH	PHB (g/L)	Biomass (g/L)	PHB % accumulation
1	5.00	6.00	3.00	10.00	5.00	100.00	6	0.207765	2.84	7.315659
2	25.00	1.00	3.00	10.00	5.00	250	8	0.843294	4.42	19.07905
3	25.00	1.00	3.00	10.00	15.00	100	6	0.261176	3.02	8.648228
4	5.00	6.00	3.00	5.00	15.00	250	8	0.05	1.12	4.464286
5	15.00	3.50	1.75	7.50	10.00	175	7	0.256941	5.46	4.705882
6	25.00	6.00	3.00	5.00	5.00	100	8	0.265765	1.32	20.13369
7	25.00	6.00	0.50	10.00	15.00	250	6	1.64706	5.4	30.50109
8	25.00	1.00	0.50	5.00	15.00	100	8	0.264471	2.24	11.80672
9	5.00	1.00	0.50	5.00	5.00	100	6	0.139294	2.12	6.570477
10	5.00	1.00	0.50	10.00	5.00	250	8	0.160424	2.44	6.574735
11	25.00	6.00	0.50	5.00	5.00	250	6	0.852706	3.96	21.53298
12	15.00	3.50	1.75	7.50	10.00	175	7	0.421647	5.64	7.476012
13	5.00	1.00	3.00	5.00	15.00	250	6	0.293647	2.82	10.41302
14	15.00	3.50	1.75	7.50	10.00	175	7	0.256941	4.98	5.159461
15	5.00	6.00	0.50	10.00	15.00	100	8	0.05	1.38	3.623188
16	15.00	3.50	1.75	7.50	10.00	175	7	0.213647	4.34	4.922743

Table 5 p value, coefficient, and standard error for the effect of medium components on growth and PHB production

Factors	p values	Coefficient	Standard error
Fructose (g/L)	0.0047 ^b	0.27	0.066
NH ₄ Cl (g/L)	0.2041 ^a	0.093	0.066
KH ₂ PO ₄ (g/L)	0.0966 ^c	-0.099	0.066
Inoculum (%)	0.1443 ^a	0.11	0.066
Trace element (mL/L)	0.9060 ^a	8.092E-0.03	0.066
Agitation speed (rpm)	0.0122 ^b	0.22	0.066
Initial pH	0.0611 ^c	-0.15	0.066

^a Nonsignificant at p < 0.05

^b Significant positive effect

^c Significant negative effect

$$\begin{split} Biomass &= +6.47 + 2.27A - 0.172B + 0.416C + 0.739D \\ &- 0.0913AB - 0.279AC + 0.389AD \\ &- 0.299BC - 0.149BD - 0.221CD + 0.204A^2 \\ &- 0.856B^2 - 0.374C^2 - 0.541D^2 \end{split}$$

In which, A is fructose, B is initial pH, C is KH₂PO₄, and D is agitation speed.

In order to determine the optimal levels of each variable for maximum PHB production, three-dimensional response surface plots were constructed by plotting the response (PHB production or biomass) on the Z-axis against any two variables which their interaction are most significant (according to their *p* values) while maintaining other variables at their optimal levels (Figs. 2, 3). Fructose and agitation speed showed significant influence on PHB accumulation either as individual or in interaction with each other (p < 0.0001). The results indicate that the concentration of the carbon source bears a direct relationship with PHB production. The 3D plot (Figs. 2, 3) clearly shows a linear increase in PHB production and biomass when the fructose concentrations were increased. The p values for the model (<0.0001) and for lack of fit (<0.0001) for biomass and PHB production also suggested that the obtained experimental data were a good fit with the model.

It was found by perturbation plot (Fig. 4) that PHB accumulation enhanced up to agitation speed of 175 rpm, but from 175 to 250, this increase is gradual and has no significant effect on PHB production. This is in contrast to previous studies, which reported agitation speed higher than 200 rpm showed negative influence on PHB content (Khosravi Darani et al. 2004). The best values for KH₂PO₄ and agitation speed as one parameter were 1.75 g/L and 175 rpm, respectively. The PHB content decreases when initial pH increases from 6 to 8 (Figs. 1, 2). This may be the reason that ammonium chloride was the best nitrogen source because it could decrease the pH of medium, although the important matter here is the effect of factors in combination with each other.

From the CCD results, it was found that PHB content increased from 1.03 g/L using Plackett-Burman design to 7.48 g/L. Thus, the optimum values of the media



Table 6 Experimental design and response of Central composite design

Experiment No.	Fructose (A) (g/L)	Initial pH (B)	KH ₂ PO ₄ (C) (g/L)	Agitation speed (D)	PHB (g/L)	Biomass (g/L)	PHB % accumulation
1	5.00	6	0.5	100	0.706975	2.72	25.99173
2	25.00	6	0.5	100	3.625725	6.38	56.82955
3	5.00	6	3	100	1.095725	3.22	34.02873
4	25.00	6	3	100	3.935725	7.24	54.36084
5	5.00	8	0.5	100	0.3301	2.7	12.22593
6	25.00	8	0.5	100	2.741975	5.96	46.00629
7	5.00	8	3	100	0.526975	3.1	16.99919
8	25.00	8	3	100	2.498225	6.42	38.91316
9	5.00	6	0.5	250	0.1726	2.72	6.345588
10	25.00	6	0.5	250	4.275725	7.76	55.09955
11	5.00	6	3	250	0.814475	4.02	20.26057
12	25.00	6	3	250	3.913225	7.94	49.28495
13	5.00	8	0.5	250	0.171975	2.74	6.27646
14	25.00	8	0.5	250	6.763225	10.38	65.15631
15	5.00	8	3	250	0.1476	3.22	4.583851
16	25.00	8	3	250	6.375725	7.1	89.79894
17	-5.00	7	1.75	175	0.0501	1.68	2.982143
18	35.00	7	1.75	175	7.481975	11.6	64.49978
19	15.00	7	-0.75	175	0.143225	2.06	6.95267
20	15.00	7	4.25	175	3.050725	6.6	46.22311
21	15.00	5	1.75	175	3.613225	4.01	90.10536
22	15.00	9	1.75	175	0.0656	1.46	4.493151
23	15.00	7	1.75	25	0.176038	1.26	13.97123
24	15.00	7	1.75	325	2.225725	6.06	36.72814
25	15.00	7	1.75	175	3.213225	6.46	49.74033
26	15.00	7	1.75	175	3.081975	6.3	48.92024
27	15.00	7	1.75	175	3.269475	6.7	48.79813
28	15.00	7	1.75	175	2.906975	6.38	45.56387
29	15.00	7	1.75	175	3.069475	6.46	47.51509
30	15.00	7	1.751	175	3.038225	6.52	46.59854



Design-Expert® Software



Fig. 2 3D surface showing the interactive effect of fructose (g/L) and agitation speed (rpm) on PHB content (g/L) after 48 h of cultivation

Fig. 3 3D surface showing the interactive effect of fructose (g/L) and agitation speed (rpm) on biomass (g/L) after 48 h of cultivation



Fig. 4 Perturbation plot showing the effect of fructose, agitation speed, initial pH, and KH_2PO_4 on PHB content after 48 h of cultivation

components (Table 6) after CCD design are as follows: fructose, 35 g/L, KH₂PO₄, 1.75 g/L, MgSO₄.7H₂O 1.2 g/ L, Citric acid 1.7 g/L, trace element 10 mL/L m initial pH = 7, and agitation speed 175. This is the characteristic of CCD, which designs some runs up or down to limitation values. In these experiments, it is found that higher value of PHB can be reached by increasing fructose concentration up to limitation. This is the case that needs more evaluation. After knowing the possible direction for maximizing PHB production, the optimization was done by "Point prediction" technique. A maximum PHB content of 7.70 g/ L was predicted that was near to what reported (7.48 g/L).

Conclusion

The methodology of Plackett–Burman was very useful for the short-listing of relevant variables for process optimization. Using the method of factorial design and response surface analysis, it was possible to determine optimal operating conditions, which led to higher concentrations and productivity without any loss of yield. Shake flask cultivation studies depicted 7.48 g/L PHB content under optimized nutrient and physical variables, which is higher than the previous report under similar condition (Tripathi et al. 2013). Optimized media obtained by response surface methodology consisted of fructose, 35 g/L; KH₂PO₄, 1.75 g/L; MgSO₄·7H₂O 1.2 g/L, Citric acid 1.7 g/L, trace element 10 mL/L, initial pH = 7, agitation speed 175, and temperature 30 °C. In this study, fructose and agitation speed showed significant effect on PHB content. However, PHB production can be controlled by precisely manipulating pH and phosphorous source.

These data obtained from batch culture would be used for simulation of nutrient feeding strategies in fed batch/ continuous cultivation for over production of PHB. The possible explanation for the rise in PHB pool with increasing amount of fructose is due to boost in growth and the availability of plenty of precursors, i.e., acetate for biosynthesis of PHB, as reported in many bacteria (Zinn et al. 2003; Galindo et al. 2007). These results show that Plackett-Burman design not only is able to optimize the nutritional constituents, but also the physical variables. On the basis of the data obtained in the present study, C. necator is capable of PHB accumulation up to 90 % of dry cell weight. PHB accumulation was induced by phosphate limitation instead of nitrogen limitation. In this connection, Cavaillé et al. (2013) reported that PHB accumulation was enhanced when growth was restricted due to unavailability of phosphorus. Using byproducts that are rich in fructose such as palm date powder or fruit juice for making the process of production cost-effective and some molecular studies are needed to increase more PHB content and making it cost-effective.

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Abbreviations

PHA	Polyhydroxyalkanoate
PHB	Polyhydroxybutyrate
CCD	Central composite design
RSM	Response surface methodology

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