

The surface adsorption of some hospital airborne microorganisms by nano-based columns

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Abstract The aim of this study was to evaluate the adsorption of some hospital airborne microorganisms, e.g. *Staphylococcus aureus*, *Streptococcus pyogenes*, *Pseudomonas aeruginosa*, and *Candida albicans* by some inorganic nanoparticles. First, each microbial suspension (2×10^4 and 2×10^6 CFU/mL) was separately added to adsorbent columns (1 and 2 g), containing CaSO_4 , CaCO_3 , and FeSO_4 nanoparticles. Then, the colony count reduction was measured for each adsorbent column. This study showed that all nanoparticles could adsorb all kinds of microorganisms. Importantly, the combination of CaSO_4 and CaCO_3 nanoparticles led to better adsorption property. Moreover, it was found that the adsorption was related to adsorbent weight and microbial density. The authors suggest that these adsorbent columns are good choice to remove hospital airborne microorganisms.

Keywords Adsorption · Airborne · Microorganisms · Adsorbent column · Nanoparticles

Introduction

Hospital-acquired infections or nosocomial infections are induced by different pathogens, e.g. *Staphylococcus aureus*, *Candida albicans*, *Pseudomonas aeruginosa*, *Streptococcus pyogenes*, found in hospital environment. These infections can be acquired by a patient during a hospital visit. Approximately, 1.7 million hospital-associated infections have been reported by the Centers for Disease Control (CDC), in USA (Kelly and Monson 2012; Lobdell et al. 2012).

Nowadays, air filtration systems are applied in hospital to remove virus, bacteria, fungi, pollen grains, and dusts. Basically, the concentration of airborne particles in a hospital is a balance between the rate of their release and the rate of their removal. Theoretically, air filters must remove particles before the reenter the hospital. Pollen grains, dusts, and fungal spores are not pathogenic, but can cause allergic rhinitis and asthma. All of them can be removed by traditional air filtration. Of airborne particles, viruses are too small, below $0.1 \mu\text{m}$, and escape from micron-size pores. Based on CDC guidelines, a hospital air filtration system can play an important role in airborne particles. The most of air filtration systems are combined with UV lamp to kill captured microbes (Karottki et al. 2013; Ma and Henderson 2013; Wichmann et al. 2013).

In the most developing and undeveloped countries, no qualified air filtration system is used in the hospitals, because of bad hospital management and high price of air filtration systems. On the other hand, the sub-micron particles cannot be removed by traditional air filtration systems. Nanoparticles have high surface area and can be conjugated by different chemicals (Jung et al. 2011). We think that nanoparticles are good choice to adsorb airborne particles. They can be used as adsorbent column to remove

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both big particles, e.g. dusts, pollen grains, and fungi, and small particles, e.g. viruses and bacteria.

Materials and methods

Materials

CaSO₄, CaCO₃, and FeSO₄ nanoparticles were provided from Zyst Fannavar Shargh Company, Iran. These nanoparticles had been characterized by scanning electron microscopy (SEM). Sabouraud dextrose agar (SDA), nutrient agar (NA), and RPMI1640 were purchased from Invitrogen, UK.

Preparation of microbial suspension

In this study, five microbial species were used, including *Staphylococcus aureus*, *Streptococcus pyogenes*, *Pseudomonas aeruginosa*, and *Candida albicans*. These isolates were obtained from Department of Microbiology, Paramedicine University, Shahid Sadoughi University of Medical Sciences, Yazd, Iran. One colony of *Staphylococcus aureus*, *Streptococcus pyogenes*, and *Pseudomonas aeruginosa* was inoculated on nutrient agar and incubated at 37 °C for 24 h. In case of *Candida albicans*, one colony was inoculated on SDA and incubated at 25 °C for 48 h. After incubation, all were washed by normal saline (NS). Finally, the density of each isolate was adjusted to 2×10^6 CFU/mL by NS.

Preparation of adsorbent column

The adsorbent column which used in this study had five layers (Fig. 1). Layers 1, 3, and 5 were cellulose, and layers 2 and 4 were adsorbent. Here, six adsorbents were prepared, including:

1. Adsorbent 1: Containing CaSO₄ nanoparticles
2. Adsorbent 2: Containing CaCO₃ nanoparticles
3. Adsorbent 3: Containing FeSO₄ nanoparticles
4. Adsorbent 4: Containing CaSO₄ and CaCO₃ nanoparticles, at 50 % w/w
5. Adsorbent 5: Containing CaSO₄ and FeSO₄ nanoparticles, at 50 % w/w
6. Adsorbent 6: Containing CaCO₃ and FeSO₄ nanoparticles, at 50 % w/w

Adsorption study

Here, three individual experiments were done for each adsorbent column, including:

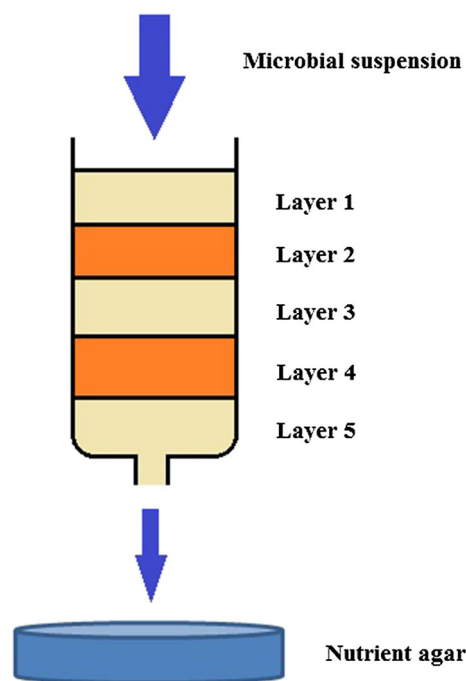


Fig. 1 Structure of adsorbent column used in this study. As seen, it had five layers. Layers 1, 3, and 5 were cellulose, and layers 2 and 4 were nanoparticles

- A. Adsorbent column had 1 g nanoparticle, and the density of microbial suspension was 2×10^4 CFU/mL
- B. Adsorbent column had 2 g nanoparticle, and the density of microbial suspension was 2×10^4 CFU/mL
- C. Adsorbent column had 1 g nanoparticle, and the density of microbial suspension was 2×10^6 CFU/mL

First, 5 mL of microbial suspension was separately added to each adsorbent column. After 5 min, the first droplet (50 µL) which passed from column was collected, and then the quantity of microbes was read by cell counter. Finally, the percentage of colony count reduction was calculated, according to Formula 1.

$$\text{Colony count reduction} = (A - B) \times 100/B \quad (1)$$

where A is CFU of adsorbent column and B is CFU of negative control. The column that had no nanoparticles was considered as negative control.

Statistical analysis

All tests were done three times, and the results are shown as the mean \pm standard deviation (SD). Parametric test (ANOVA) was applied to detect the significant difference. This test was carried out by SPSS software (V.16.0 for



Windows; SPSS Inc., USA), and $P < 0.05$ was considered as a significant difference.

Results and discussion

Figure 2 shows the SEM image of CaSO_4 , CaCO_3 , and FeSO_4 nanoparticles. As seen, all had distribution size of 50 ± 10 nm, approximately. Figures 3a, b, 4a, b, and 5a, b show the colony count reduction of microbial suspension when passed from adsorbent column 1, 2, 3, 4, 5, and 6, respectively. Here, the density of microbial suspension was 2×10^4 CFU/mL, and the weight of adsorbent was variable, 1 and 2 g. As the first finding, all adsorbent columns could adsorb all kinds of microorganisms which studied in this article. As the second finding, in all treated groups, adsorbent weight of 2 g was better than adsorbent weight of 1 g, i.e. the more colony count reduction, the more adsorbent weight. As the third finding, the highest and least colony count reduction were achieved when microbial suspension was passed from adsorbent columns 4 and 3, respectively.

Figures 6a, b, 7a, b, and 8a, b show the colony count reduction of microbial suspension when passed from adsorbent columns 1, 2, 3, 4, 5, and 6, respectively. Here, the weight of adsorbent was 1 g, and the density of microbial suspension was variable, 2×10^4 and 2×10^6 CFU/mL. As an important result, the decrease in colony count reduction was observed when the density of microbial suspension was increased. Here, the highest and least colony count reductions were also achieved when microbial suspension was passed from adsorbent columns 4 and 3, respectively.

Staphylococcus aureus, *Streptococcus pyogenes*, *Pseudomonas aeruginosa*, and *Candida albicans* are the main cause of nosocomial infections (Gomes et al. 2014). These agents can spread in outdoor and outdoor air of hospital and lead to skin and lung infections (Pourakbari et al. 2015). Nowadays, some disinfectant agents are used to kill or inhibit nosocomial agents at hospital surfaces (Dancer

2014). But, it must be mentioned that these agents can be found in air, and surface disinfectant cannot damage them. The authors propose that the use of adsorbent materials is a key factor to remove them. Nanoparticles have a high surface area and can be attached to surface molecules of pathogens by non-covalent bindings (Daniel et al. 2014). In this study, we showed the adsorption property of CaSO_4 , CaCO_3 , and FeSO_4 nanoparticles. Among them, CaCO_3 nanoparticles had highest adsorption property, and CaSO_4 nanoparticles had in the later level. This study showed that the combination of CaSO_4 and CaCO_3 nanoparticles led to better adsorption property. It must be noted the synergetic effect was also seen in the combination of CaSO_4 and FeSO_4 nanoparticles, and CaCO_3 and FeSO_4 nanoparticles. The aim of this study was to investigate the efficacy of CaSO_4 , CaCO_3 , and FeSO_4 nanoparticles to adsorb *Staphylococcus aureus*, *Streptococcus pyogenes*, *Pseudomonas aeruginosa*, and *Candida albicans*. Here, six adsorbent columns were used, including CaSO_4 nanoparticles, CaCO_3 nanoparticles, FeSO_4 nanoparticles, ($\text{CaSO}_4 + \text{CaCO}_3$) nanoparticles, (CaSO_4 and FeSO_4) nanoparticles, and (CaCO_3 and FeSO_4) nanoparticles. We found that all of them could adsorb microbes, i.e. all microorganisms which studied could be adsorbed by adsorbent column. Moreover, the best result was for adsorbent column 4, with weight of 2 g, and microbial suspension of 2×10^6 CFU/mL. As mentioned, adsorbent column 4 had CaSO_4 and CaCO_3 nanoparticles, at 50 % w/w. The surface adsorption of microbes by nanoparticles is an important finding. This phenomenon can be applied in manufacturing of adsorbent sheet or filter to remove microbes in important places, e.g. ICU and CCU. Nosocomial infections can be occurred in hospitals by both pathogenic and opportunistic microbes. It is important to control the quantity of microbes in air of hospitals.

Lukasik et al. investigated the effects of mono-, di-, and trivalent salts (NaCl , MgCl_2 , and AlCl_3) on the adsorption of several viruses (MS2, PRD-1, fX174, and poliovirus 1) to microporous filters at different pH values. The increase in adsorption was observed when AlCl_3 and HCl were added to

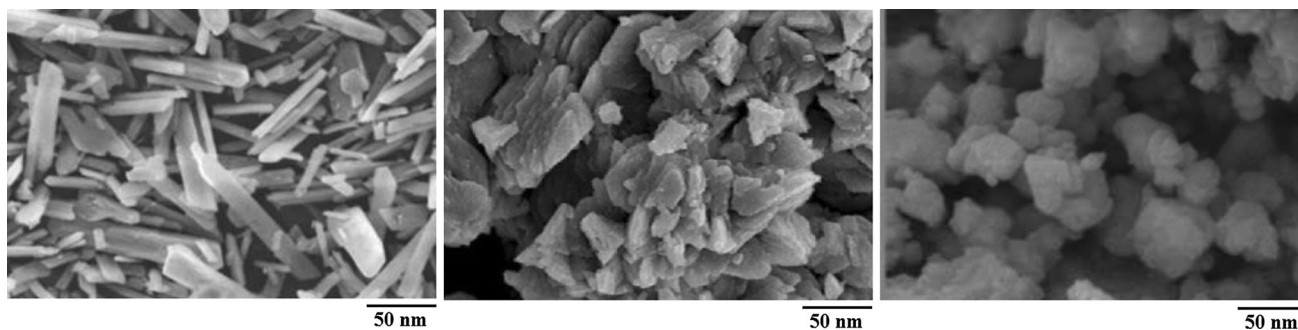


Fig. 2 SEM image of CaSO_4 , CaCO_3 , and FeSO_4 nanoparticles



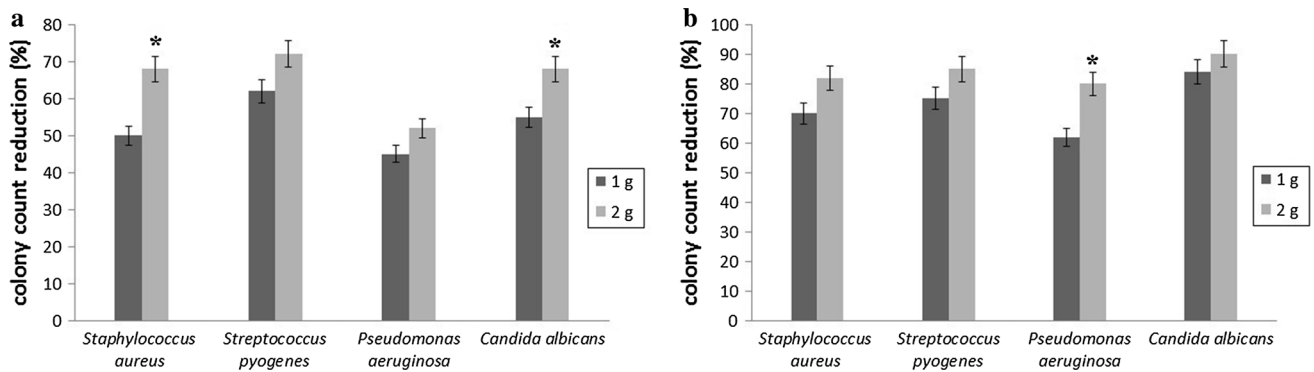


Fig. 3 Colony count reduction when microbial suspension was passed from adsorbent column 1 (a), 2 (b). Adsorbents 1 and 2 had CaSO₄ and CaCO₃ nanoparticles, respectively. Here, the density of

microbial suspension was 2×10^4 , and the content of nanoparticles was variable, 1 and 2 g. * $P < 0.05$ compared when the adsorbent had 1 g nanoparticles. $n = 3$

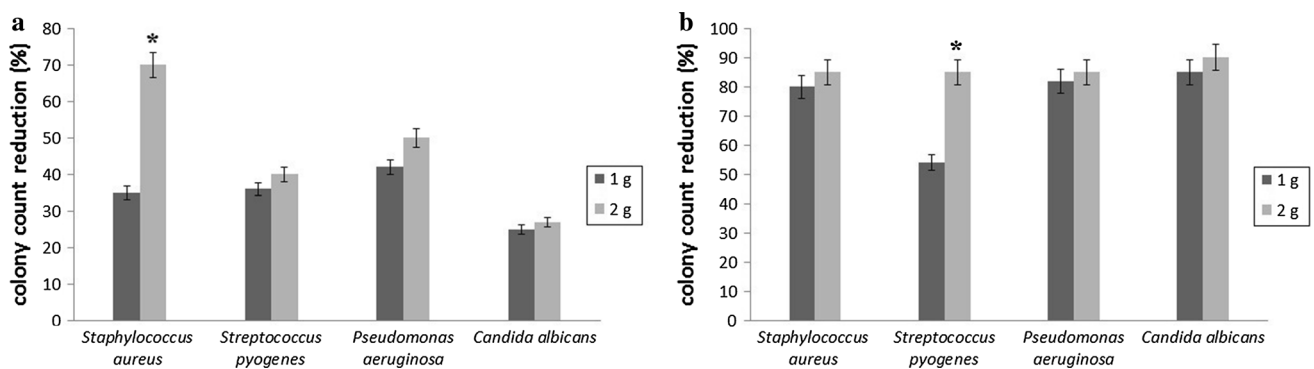


Fig. 4 Colony count reduction when microbial suspension was passed from adsorbent column 3 (a), 4 (b). Adsorbent 3 had FeSO₄ nanoparticles, and adsorbent 4 had CaSO₄ and CaCO₃ nanoparticles.

Here, the density of microbial suspension was 2×10^4 , and the content of nanoparticles was variable, 1 and 2 g. * $P < 0.05$ compared when the adsorbent had 1 g nanoparticles. $n = 3$

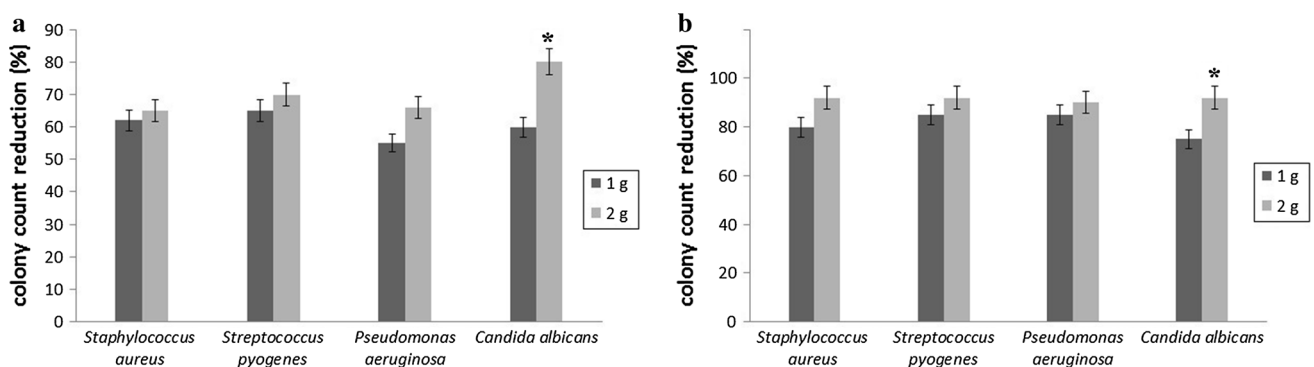


Fig. 5 Colony count reduction when microbial suspension was passed from adsorbent column 5 (a), 6 (b). Adsorbent 5 had CaSO₄ and FeSO₄ nanoparticles, and adsorbent 6 had CaCO₃ and FeSO₄

nanoparticles. Here, the density of microbial suspension was 2×10^4 , and the content of nanoparticles was variable, 1 and 2 g. * $P < 0.05$ compared when the adsorbent had 1 g nanoparticles. $n = 3$

water which was resulted in the decrease in the pH of the water (Lukasik et al. 2000). Verdenelli et al. studied the activity of biostatic agents on the microbial colonization of panel filters. They showed the integrity of the filters and the lower release of microorganisms from treated filters

(Verdenelli et al. 2003). Miśkiewicz-Peska et al. used antimicrobial air filter treatment. They examined woven air filters made of polypropylene. It was found that antibacterial filter treatment resulted in an evident reduction in living bacterial cells (Miśkiewicz-Peska and Łebkowska 2011).



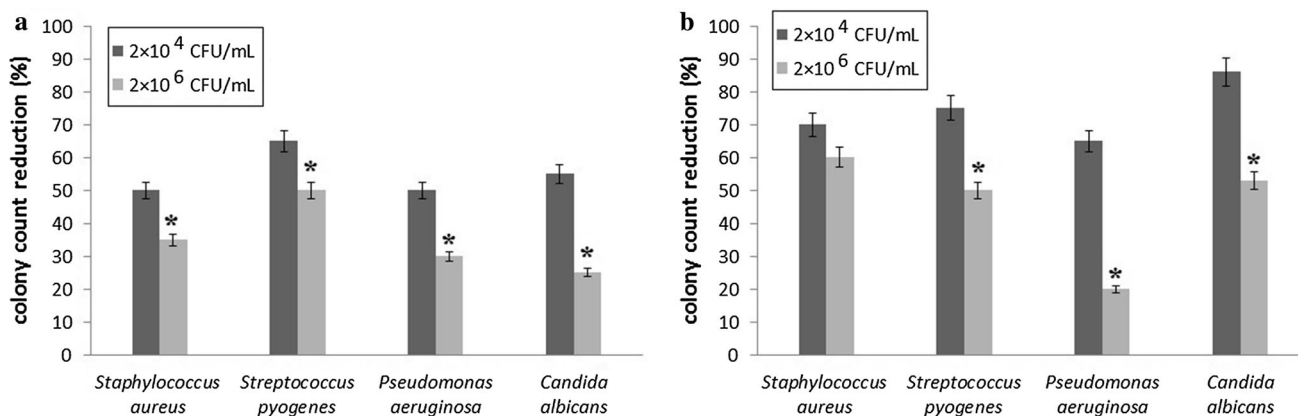


Fig. 6 Colony count reduction when microbial suspension was passed from adsorbent column 1 (a), 2 (b). Adsorbents 1 and 2 had CaSO₄ and CaCO₃ nanoparticles, respectively. Here, the content of

nanoparticles was 1 g, and the density of microbial suspension was variable, 2×10^4 and 2×10^6 CFU/mL. * $P < 0.05$ compared when the microbial suspension was 2×10^4 , $n = 3$

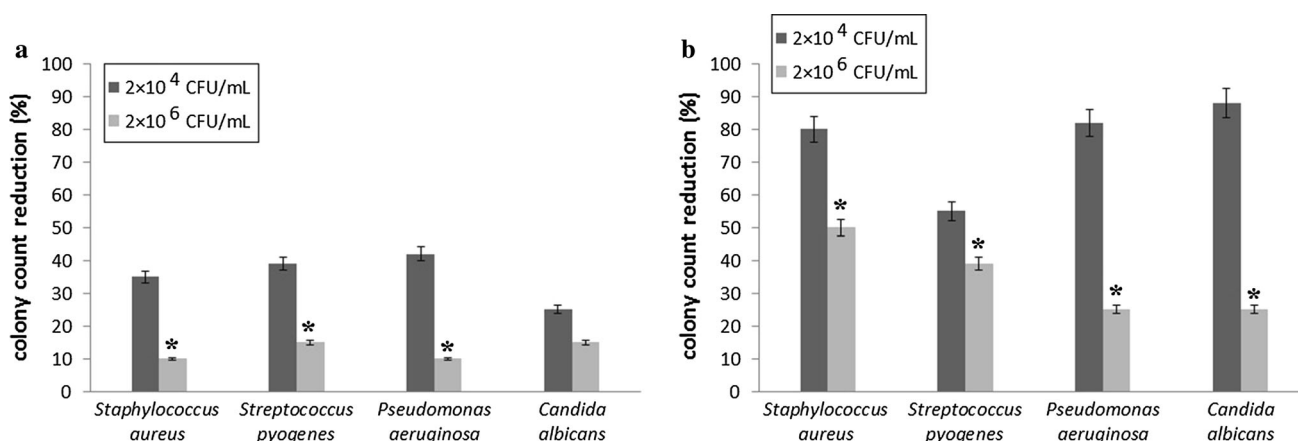


Fig. 7 Colony count reduction when microbial suspension was passed from adsorbent column 3 (a), 4 (b). Adsorbent 3 had FeSO₄ nanoparticles, and adsorbent 4 had CaSO₄ and CaCO₃ nanoparticles. Here, the content of nanoparticles was 1 g, and the density of

microbial suspension was variable, 2×10^4 and 2×10^6 CFU/mL. * $P < 0.05$ compared when the microbial suspension was 2×10^4 , $n = 3$

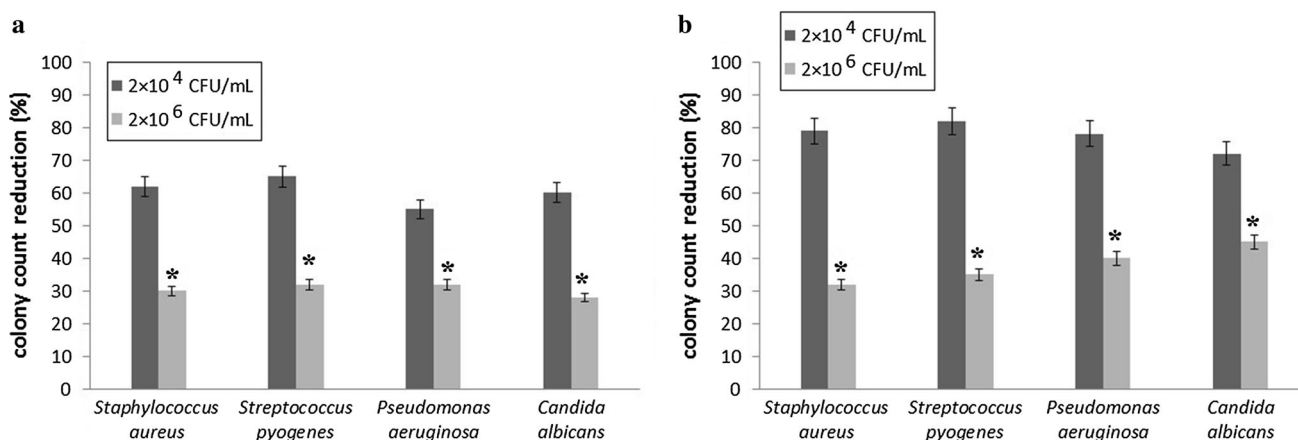


Fig. 8 Colony count reduction when microbial suspension was passed from adsorbent column 5 (a), 6 (b). Adsorbent 5 had CaSO₄ and FeSO₄ nanoparticles, and Adsorbent 6 had CaCO₃ and FeSO₄ nanoparticles. Here, the content of nanoparticles was 1 g, and the

density of microbial suspension was variable, 2×10^4 and 2×10^6 CFU/mL. * $P < 0.05$ compared when the microbial suspension was 2×10^4 , $n = 3$



Conclusion

It can be concluded that all nano-based columns which investigated in this study could adsorb *Staphylococcus aureus*, *Streptococcus pyogenes*, *Pseudomonas aeruginosa*, and *Candida albicans*. Moreover, the combination of CaSO₄ and CaCO₃ nanoparticles led to better adsorption property. These adsorbent columns are good choice to remove hospital airborne microorganisms.

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Compliance with ethical standards

Conflict of interest There is no conflict of interest to declare.

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